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 Limiting Reagent 1.  $N_2 + 3H_2 \rightarrow 2NH_3$  How many grams of  $NH_3$  can be produced from the reaction of 28g of  $N_2$  and  
 25 g of  $H_2$ ? 2. How many grams of the excess reagent in Problem 1 is left over? ...* Chemistry If8766 Page 66  
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 excess reagent? of the two answers is the Notes #3 to complete the 'following Limiting Reagent word problems. for  
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 limiting reagent if 79.6 grams of  $NO$  were reacted with 59.5 grams of  $O_2$ ?  $2NO + O_2 \rightarrow 2NO_2$  ... To answer this question,  
 calculate first the theoretical yield of  $N_2CH_4O$  that should be made. Then use the ... *Stoichiometry* Author: Susan Arena  
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 answer the following questions: 1) Write the balanced equation for the reaction of lead (II) nitrate with sodium

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